

1. Electronic what ?!

Definition

An electronic nose (eNose) is a device which is composed of **chemical sensors** and a **pattern-recognition system** for recognising odours [1].

Machine Learning stage

It produces a signature when an odour is introduced. ML algorithms are then used to identify the odour.

However, our eNose outputs up to **~100 time series**, from which we need to **extract a smaller set of relevant features**.

The solution proposed here is to extract them from a **physical model**.

Applications

Biomedical engineering, Food-processing industry, Mine-clearing, Cosmetics, Olfactive navigation...

2. NeOse

Our work is based on the NeOse eNose, developed by Aryballe Technologies, a French start-up located at CEA in Grenoble:



Fig. 1: NeOse device.

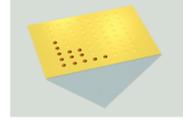


Fig. 2: Chemical sensors are peptides (small proteins) which are dropped on the golden surface of a prism.

Imaging technique :

A light beam is sent to and reflected by the surface. When interactions occur, the refraction index changes and thus the reflected light too (fig. 3).

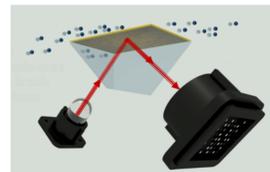


Fig. 3: Imaging technique principle.

3. Data acquisition

① Baseline acquisition

② Gas on at $t = t_s$

③ Gas off at $t = t_e$

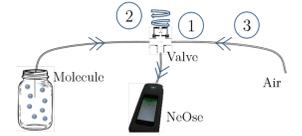


Fig. 4: Acquisition plan.

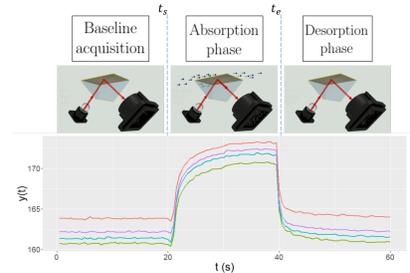
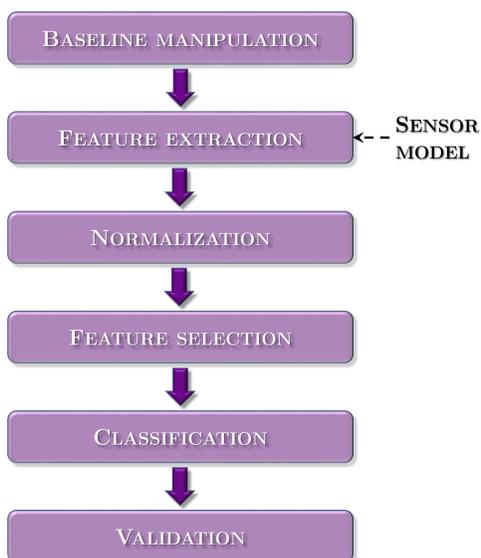


Fig. 5: Each line corresponds to the temporal response $y(t)$ of one peptide during an experiment as described above.

4. Data processing

Data are usually processed according to the plan [2]:



5. Sensor model

Molecular interaction

For each peptide \mathcal{P} , the following **binding reaction** occurs when a molecule \mathcal{M} is introduced:

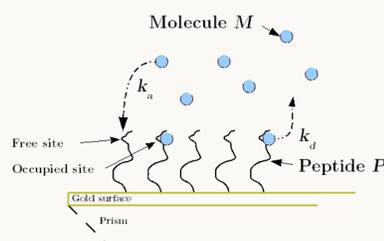
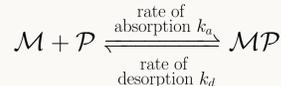


Fig. 7: Simplified molecular interaction.

Langmuir model

From this 1st order reaction, the kinetic of complex formed \mathcal{MP} is obtained using the **kinetics parameters** k_a and k_d :

$$\frac{d[\mathcal{MP}](t)}{dt} = \underbrace{k_a[\mathcal{M}](t)[\mathcal{P}]}_{\text{change in \# of occupied sites}} - \underbrace{k_d[\mathcal{MP}](t)}_{\text{Number of sites becoming free}}$$

$$\Leftrightarrow \frac{d\theta(t)}{dt} = k_a[\mathcal{M}](t) - k_d\theta(t), \quad \theta(t) = \frac{[\mathcal{MP}](t)}{[\mathcal{P}]}$$

Under some assumptions, Langmuir model arises [3]:

$$\theta(t, k_a, k_d) = \begin{cases} 0 & \text{if } t \leq t_s \\ \frac{k_a}{k_a + k_d} (1 - e^{-(k_a + k_d)(t - t_s)}) = \theta_a(t) & \text{if } t \in [t_s, t_e[\\ \theta_a(t_e) e^{-k_d(t - t_e)} & \text{if } t \geq t_e \end{cases}$$

The imaging technique doesn't measure directly the fraction of occupied site θ but a linear transformation of it:

$$y(t, \alpha, k_a, k_d) = \alpha \theta(t, k_a, k_d)$$

where α depends on molecule \mathcal{M} mass, peptide \mathcal{P} concentration and temperature \rightarrow thus, **not on the interaction!**

6. Results

Model fitting

The Langmuir model is fitted using nonlinear least-squares (with BFGS).

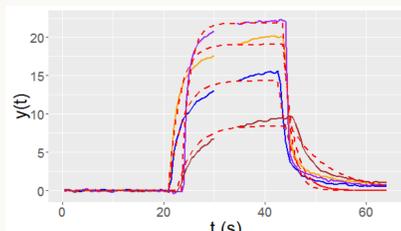
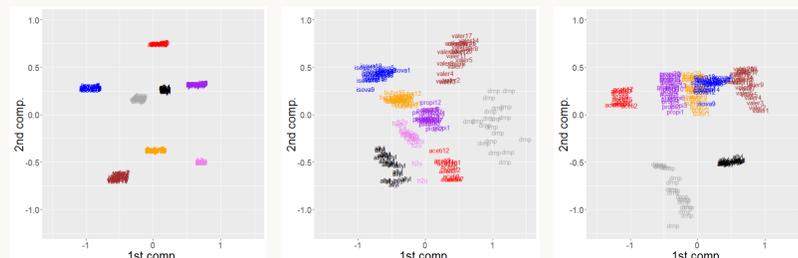


Fig. 8: Solid lines are real data after baseline subtraction (**butyric-acid**, **isovaleric-acid**, **propionic-acid**, **valeric-acid**) and dashed lines the fitted values.

Multi-Dimensional Scaling and Classification

Data set : 8 molecules, repeated 20 times at constant temperature.



(a) MDS with k_a (b) MDS with k_d (c) MDS with θ_{eq}

Fig. 9: MDS results (**acetic-acid** $\text{C}_2\text{H}_4\text{O}_2$, **propionic-acid** $\text{C}_3\text{H}_6\text{O}_2$, **butyric-acid** $\text{C}_4\text{H}_8\text{O}_2$, **valeric-acid** $\text{C}_5\text{H}_{10}\text{O}_2$, **isovaleric-acid** $\text{C}_5\text{H}_{10}\text{O}_2$, dimethylpropane, allyl-hexanoate, water).

We use the features (k_a , k_d and θ_{eq} , the static response) in a classifier such as **k-NN** and **SVM**. Each feature gives a **cross-validated score of 100%**.

7. Discussion

Conclusion

- Langmuir model explains well the absorption but is sometimes poor in the desorption phase.
- The features extracted from the dynamics are reproducible and discriminative.

Perspectives

- Explain desorption patterns.
- Collect larger and harder data set.

Acknowledgments

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[1] J. W. Gardner and P. N. Bartlett, "A brief history of electronic noses," Sensors and Actuators B: Chemical, vol. 18, no. 1, pp. 210-211, Mar. 1994.

[2] R. Gutierrez-Osuna, "Pattern analysis for machine olfaction: a review," IEEE Sensors journal, vol. 2, no. 3, pp. 189-202, 2002.

[3] Y. Hou et al., "Continuous Evolution Profiles for Electronic-Tongue-Based Analysis," Angew. Chem. Int. Ed., vol. 51, no. 41, pp. 10394-10398, Oct. 2012.